



GASIFICATION TO PRODUCE ELECTRICAL ENERGY FROM BIOSOLIDS

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Presentation Content

- Energy consumption concerns
- Biosolids as an energy source
- History of gasification
- Gasification technology
- Wastewater Residuals
- Full Scale Test Results
- Conceptual Design

Energy Consumption

- The energy required for treatment and delivery of drinking water accounts for as much as 80% of its cost, and an insufficient supply of affordable energy will have a negative effect on the price and availability of water.

Energy Consumption

- Wastewater treatment plants use 3% of all the energy generated in the US
 - Energy costs represent over 20% of a treatment plants operating budget
- Wastewater facilities estimated in the top 10 industries for emissions of GHGs.

Energy Consumption

- The energy usages commonly reported and observed in the field are as follows:
 - Aeration 35 to 75% (typically at 50%);
 - Wastewater pumping (influent) 10 to 25% (typically at 15%); and
 - Facilities (lighting and heating, ventilation, and air conditioning) 5 to 15% (typically at 10%).

Energy Consumption

- The typical unit electricity requirements given for publicly owned treatment works are:
 - 908 kJ/m³ (955 kWh/mil. gal) for trickling filter plants,
 - 1257 kJ/m³ (1322 kWh/mil. gal) for activated sludge plants,
 - 1466 kJ/m³ (1541 kWh/ mil. gal) for advanced wastewater treatment (with filtration), and
 - 1817 kJ/m³ (1911 kWh/mil. gal) for advanced treatment (with filtration and nitrification).

Renewable Energy

- Energy production from biosolids is truly is a sustainable energy source.
 - Anaerobic Digestion
 - Gasification
- CHP to augment energy production
 - Thermal and Electrical

Anaerobic Digester Gas

- Heating digesters
- Heating buildings
- As a renewable energy source for generating electricity
 - About 550 to 800 kWh per dry ton of biosolids

Disadvantages

- Anaerobic digestion is a biological process
 - Subject to toxic discharges
 - Subject to biological imbalances
- Although about 45% of the biological solids are converted to biogas, there is still remaining a considerable amount for disposal.
- High concentration of ammonia back to treatment plant.

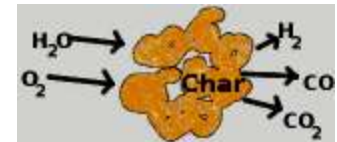
Gasification

- A process which converts carbon-containing fuels into gas.
 - The gas produced can be referred to as producer gas, synthesis gas or syngas.
 - Gasification has now been brought up to modern standards for reliability, emissions, automation and safety using today's technology



Basics of Gasification

- Gasification is a thermo-chemical reaction with the following distinct stages:
 - Drying
 - Pyrolysis (thermal decomposition)
 - Char combustion
 - Ash removal (15-20% depending on volatile solids content).



Basics of Gasification

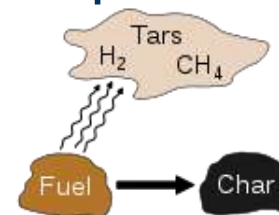
- Different gasification technologies result in different levels of particulate matter and tar in the syngas stream
 - This will affect the final application and emissions treatment

Gasification

- Performed at sub-stoichiometric oxygen concentrations and pressure (low or high)
 - Produces CO and H₂ with some methane which is a synthetic gas known as Syngas
- Syngas can be used as a gaseous fuel or for other purposes, e.g., Fischer-Tropsch liquids; methanol, etc.

What happens in a gasifier?

- Carbonaceous materials under different processes
 - Pyrolysis(or devolatilization) process occurs as the carbonaceous particle heats up.
 - Volatiles are released and char is produced
- Process is dependent on
 - properties of the carbonaceous material which
 - determines the structure and composition of the char



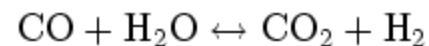
What happens in a gasifier?

- The next step is combustion
 - occurs as the volatile products and some of the char reacts with oxygen
 - form carbon dioxide and carbon monoxide
 - which provides heat for the subsequent gasification reactions



What happens in a gasifier?

- Gasification occurs as the char reacts with carbon dioxide and steam to produce carbon monoxide and hydrogen $C + H_2O \rightarrow H_2 + CO$
- Finally, reversible gas phase water gas shift reaction reaches equilibrium balancing the concentration of carbon monoxide, steam, carbon dioxide and hydrogen.



What happens in a gasifier? Summary

- limited amount of oxygen or air is introduced into the reactor to allow some of the organic material to be "burned" to produce carbon monoxide and energy
- which drives a second reaction that converts further organic material to hydrogen and additional carbon monoxide.
- Then carbon monoxide and residual water from the organic material form methane and excess carbon dioxide.

Composition of Syngas

- Typical constituents of Syngas produced from biomass gasification
 - Water vapor (from moisture content of the fuel)
 - Hydrogen
 - Carbon Monoxide
 - Methane
 - Nitrogen

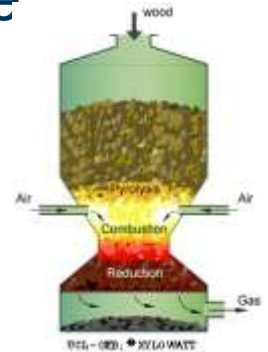
Composition of Syngas

- Syngas is a low energy gas
 - Natural gas has a lower heating value of ~1000 Btu/scf
 - Syngas typically has a lower heating value of 100-300 Btu/scf

Types of gasifiers

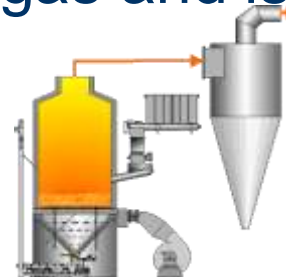
- Downdraft (small systems such as home use)

- Fuel and air flow co-currently down through the gasification vessel
- Ash discharges from the bottom of the vessel



- Fluidized Bed (large scale systems)

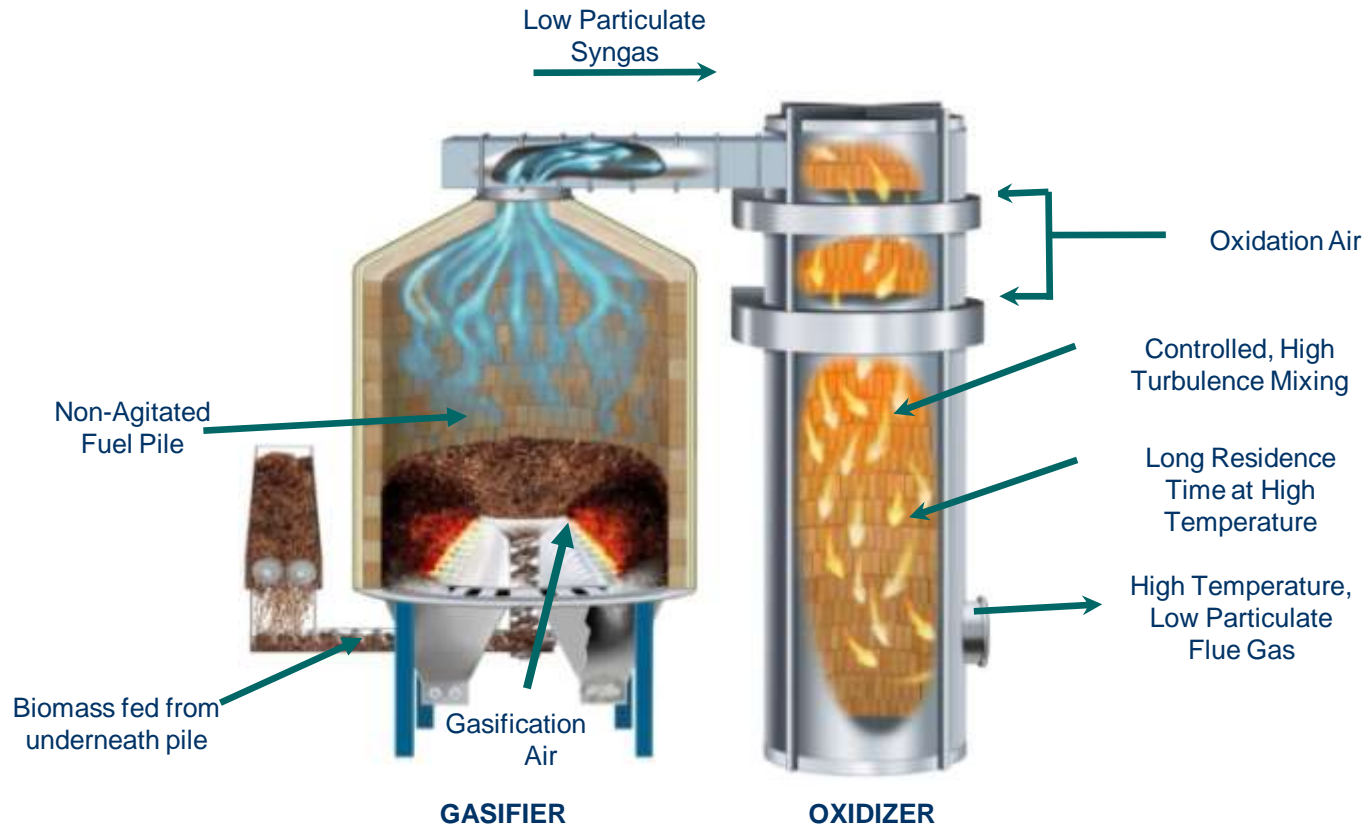
- Air fluidizes fuel bed
- Ash is entrained in the syngas and is separated from the syngas in a cyclone



Updraft Gasifier

- Updraft
 - Air flows upwards through the fuel pile, counter-current to the fuel flow
 - Ash discharges from the bottom of the vessel
 - Updraft best suits medium scale systems 1 to 10 megawatts

Updraft Gasifier



Wastewater Residuals

- Wastewater residuals (sludge, biosolids) are a renewable energy source with a relatively high BTU value.
- Characterization
 - Proximate Analysis
 - Ultimate Analysis



Biosolids Characterization

- Proximate Analysis
 - moisture content,
 - volatile content (when heated to 950 C),
 - free carbon remaining at that point,
 - ash (mineral) in the sample and
 - high heating value (HHV) based on the complete combustion of the sample to carbon dioxide and liquid water.

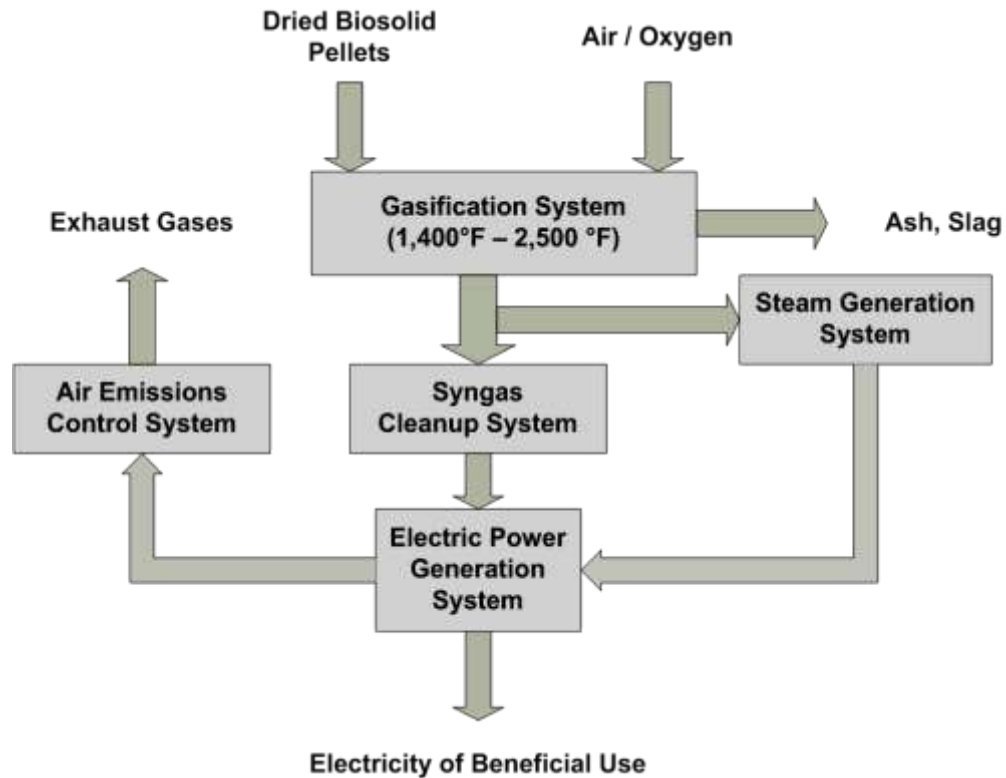
Biosolids Characterization

- Ultimate Analysis
 - composition of the biomass in wt% of carbon,
 - hydrogen and
 - oxygen (the major components)
 - as well as sulfur and nitrogen (if any).
- Fusion Point
 - Temperature at which material will vitrify or fuse

Biosolids Characteristics- primary/WAS

<i>Proximate Analysis</i>	
Moisture, %	6.7
Ash, %	16
Volatile, %	73.9
Fixed C, %	10.1
BTU/lb (HHV)	8,529
MMF BTU/lb	10,322
<i>Ultimate Analysis</i>	
Carbon, %	47.1
Hydrogen, %	6.1
Nitrogen, %	6.6
Oxygen, %	23.4
Sulfur, %	0.7
<i>Fusion Temperature</i>	
Reducing Atmosphere, C	1093
Oxidizing Atmosphere, C	1142

Gasification Waste to Energy



Full Scale Test Results

- Downdraft Gasifier
- Fluidized Bed Gasifier
- Updraft Gasifier

Downdraft Gasifier

- Throughput- @300 kg/hr
 - Originally designed for waste wood
- Design Output
 - 50 kw to grid
- Feed
 - heat dried undigested primary/WAS



Operation

- Operating temperature-900 to 950 C
 - Some clinker production
 - Did not have good method of measuring bed temperature
- Small amount of tar formation
- Small amount of ammonia and organosilicates (siloxanes)

Syngas Characteristics-Downdraft System

<i>Constituent</i>	
Hydrogen (vol %)	13.9
Carbon Monoxide (vol %)	14.4
Methane (vol %)	2.9
Carbon Dioxide (vol %)	15.4
Nitrogen (vol %)	51.9
C _x H _y (vol %)	1.4
Other (vol %)	0.1

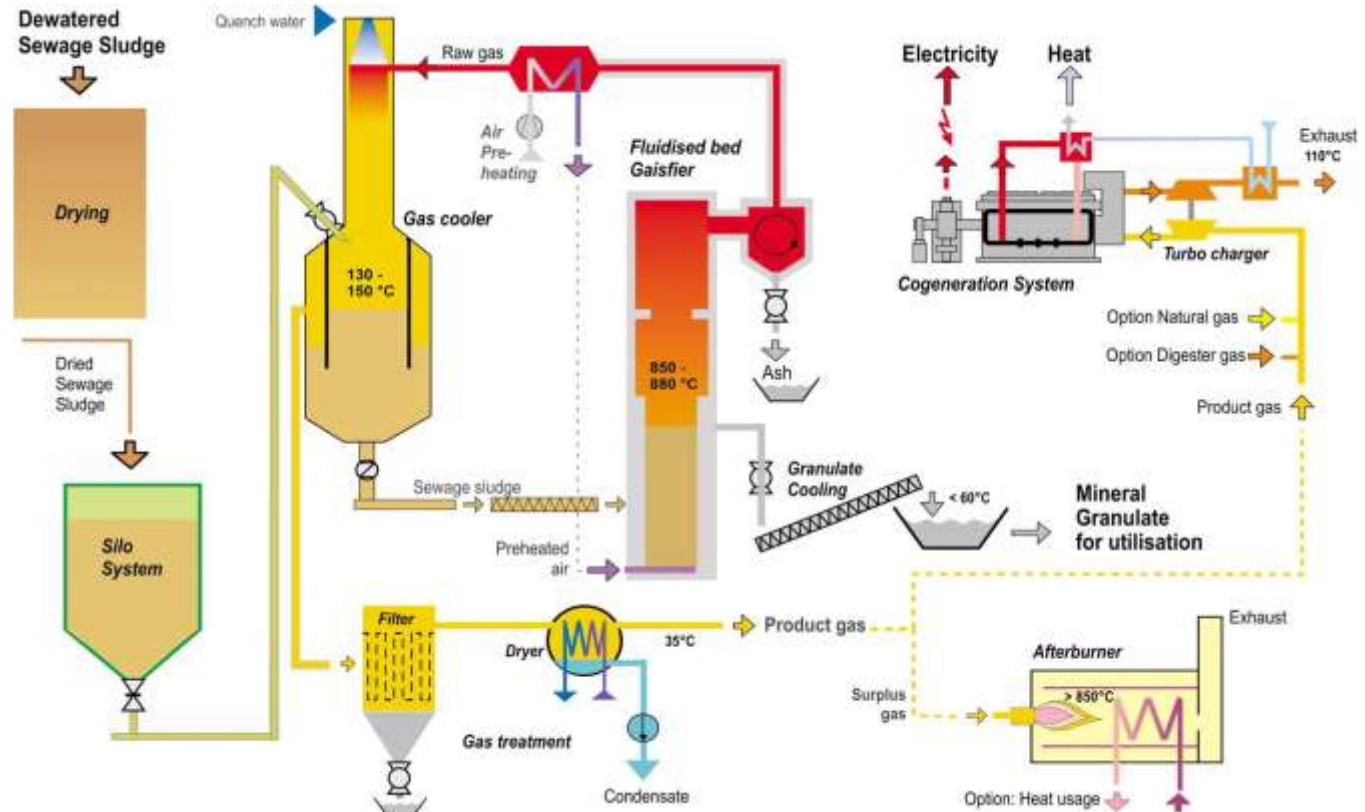
**Average Syngas Heating Value: 146 Btu/scf
Supplied 50 kw for 6.5 hrs to the electrical grid**

Fluidized Bed Gasifier

- Throughput- @ 700 kgs/hr
 - Designed for biosolids
- Output-
 - 700 kw
- Feed
 - heat dried undigested primary/WAS



Fluidized Bed Gasifier Schematic



Results

- Hydrogen 16.9 %
- Carbon Monoxide 12.8 %
- Fixed carbon in ash <8 %
- Air testing results not available at this date
- Low tar concentration in gas

Updraft Gasifier to Steam

- Throughput- @ 2000 lbs/hr
 - Designed for all types of biomass
- Output-
 - 1 megawatt
- Feed
 - Heat dried, undigested primary/WAS



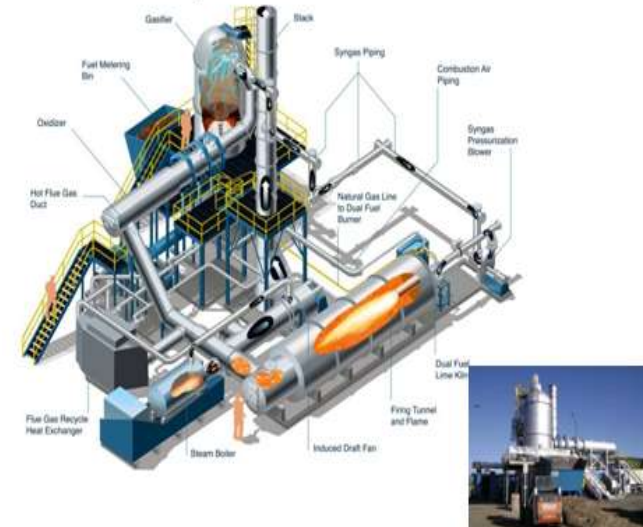
Results

- Hydrogen 15.5 %
- Carbon Monoxide 10.8 %
- Fixed carbon in ash 22 %
- Air testing excellent and within permit limits
- Tar not an issue with their technology

Updraft Gasifier to Electrical Energy

- Throughput @ 1000 lbs/hr
- Output-500 Kw
 - Designed for wood
- Feed
 - Heat dried primary/WAS

Product Development Center



Syngas Quality Dry Biosolids

<i>Syngas Composition-biosolids</i>	
Moisture	7.0
CO (mole %)	14.1
H₂ (mole %)	12.0
CH₄ (mole %)	2.9
C₂H₆ (mole %)	0.7
CO₂ (mole %)	16.6
O₂ (mole %)	0.0
N₂ (mole %)	53.6
HHV of syngas (BTU/sft³)	134.2
LHV of syngas (BTU/sft³)	123.6

Ash Discharge

- Ash was free flowing
- Visual indications are that it is low in carbon
- Ash can be used beneficially
 - Soil conditioner
 - In concrete

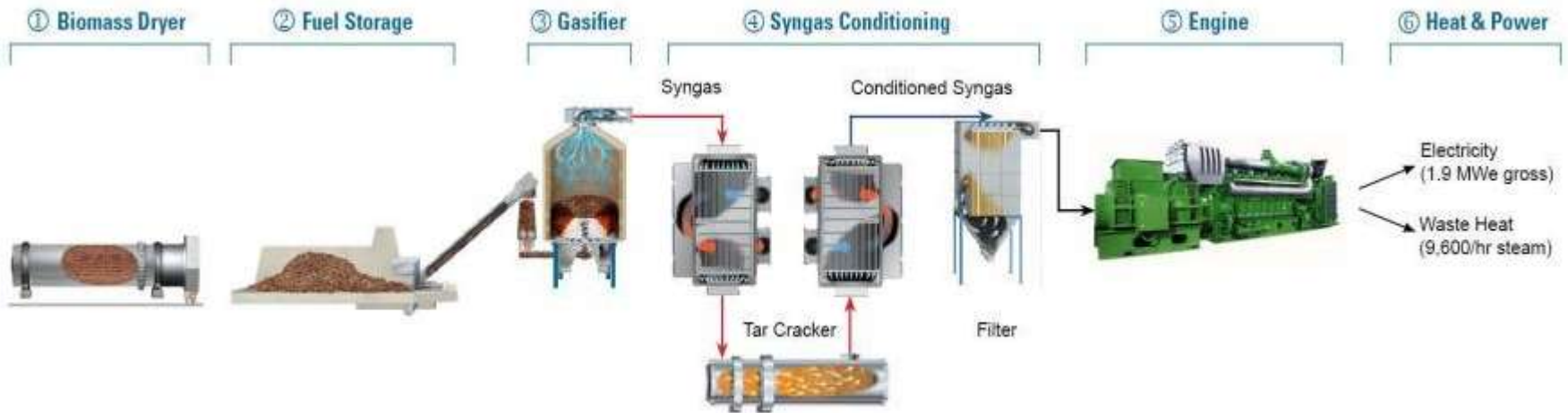


Exhaust Gases

- Operation with dry biosolids resulted in visually clean stack emissions



Conceptual System Design



- Economic at small-scale 2 – 10 MWe
- Game changing, breakthrough technology for biomass to power
- Combines technologies with IC gas engines
- Significantly more efficient than conventional steam power generation
- Firm, base load green energy vs intermittent power such as wind or solar
- No steam engines and natural gas comparable emissions for PM

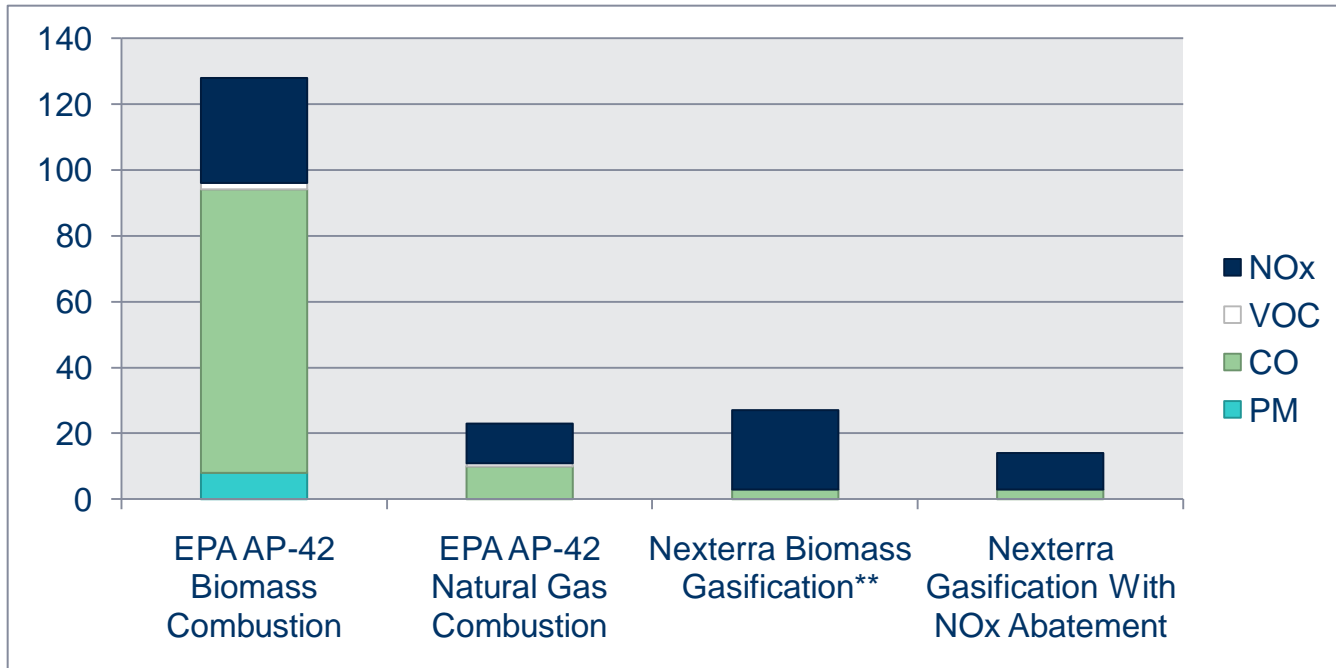
System design using Nexterra Gasifier and GE Jenbacher Engine

Internal Combustion Engine



Photo Credit GE Jenbacher Engine Division

Syngas Emissions



* Data is for on 24 MMBtu/hr system, 45% mc fuel

** Third party test results from plant at the University of South Carolina (USC)

Data Provide by Nexterra, Vancouver, BC

Tars

- Removed by Cracking
- Removed by water

Tar- (By-products of Gasification)

- Soluble tars

- Phenols
- Acetone
- Napthalene
- Pyrene
- BTEX

- Insoluble tars

- Acenaphthylene
- Fluorene
- Phenanthrene
- Pyrene
- Chrysens

Uses of Syngas

- Generate power
- Steam to heat buildings
- Liquid fuels
 - Methanol
 - Gasoline

Conclusions

- Gasification has tremendous potential
 - Need to demonstrate technology
 - Perfect tar removal system
- Biosolids are an ever renewable energy source
- Gasification will change the way we manage biosolids